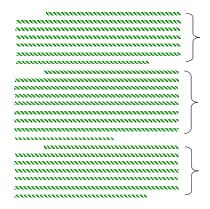
Things to keep in mind so that your scientific writing is logically structured, precise, and concise:

(1). Use paragraphs to maintain a logical structure

Every paragraph should contain roughly one idea + supporting evidence for that idea, if possible, and *this idea should be presented as concisely as possible*



paragraph 1

Idea 1, supporting examples/references, logical transition to next idea

paragraph 2

Idea 2, supporting examples/references, logical transition to next idea

etc., etc.

1

Celia's foolproof, four-step SEES* method to crank out science writing:

- 1. State the topic sentence first
- 2. Explain it
- 3. Give an <u>e</u>xample, <u>e</u>xpand, or present <u>e</u>vidence
- 4. <u>Summarize</u> it in a way that leads logically to the next topic sentence

*State → Explain → Exemplify → Summarize

Tip: Use the same construction paradigm for paragraphs, subsections, and sections of your paper

2

Writing Workshop #1: Paragraph Structure

The high-pressure and temperature phase transition of dioxides is of fundamental interest in solid-state physics, chemistry, and geosciences. In many dioxides, TiO2 is well known as an important wide-gap oxide semiconductor with various industrial applications such as electrochemical solar cells and photocatalyst due to the characteristic high refractive index [1-9]. Apart from those technological aspects, high-pressure transformations of TiO₂ have attracted special attention as a low-pressure analog of SiO2, the most abundant component of the Earth's mantle. A number of experimental and theoretical studies have revealed many crystalline polymorphs of TiO₂ at high pressures and high temperatures [10-14]. At ambient conditions, rutile is the most stable phase of TiO2. Anatase and brookite are also known as natural minerals. All of these phases transform to an α-PbO₂-type, to an orthorhombic-I-type, then finally to a cotunnite-type structure at approximately 50 GPa [11,14]. The cotunnite-type polymorph is identified as the highest-pressure phase, as in many dioxides [15]. Although the analogy of the phase change to the cotunnite structure was applied to TiO₂ [16,17], a very recent ab initio study predicted a different phase transition from the pyrite-type structure to an unexpected Fe₂P-type structure (hexagonal, space group P62m) (Fig. 1) at 690 GPa, bypassing the cotunnite-type phase stability at low temperature [18]. Since no dioxides or difluorides with this crystal structure were reported, physical and chemical properties of this new class of oxide are still unknown. Although the extremely high transition pressure predicted in SiO₂ seems unreachable in the laboratory, TiO₂ shows significantly lower transition pressures. For instance, the α-PbO₂ phase stabilizes at ~10 Gpa in TiO₂, while the same phase at 100 GPa in SiO₂. High-pressure behavior of TiO₂ is therefore a key to understanding the rich polymorphism in the metal dioxide systems, in particular, the post-cotunnite phase relations. However, all the studies performed on TiO₂ were limited below 100 GPa, and no post-cotunnite phase has been identified. In this study, we investigate the applicability of the Fe₂P-type structure to TiO₂ both theoretically and experimentally.

3

Topic sentences

The high-pressure and -temperature phase transitions of dioxides are of fundamental interest in solid-state physics, chemistry, and the geosciences. In particular, because of its high refractive index [1-9], TiO₂ is an important wide-gap oxide semiconductor for various industrial applications, such as electrochemical solar cells and photocatalysts. Additionally, high-pressure transformations of TiO₂ have attracted special attention as a low-pressure analog of SiO₂, the most abundant component of the Earth's mantle.

Previous experimental and theoretical studies reveal many crystalline polymorphs of TiO₂ at high pressures and high temperatures [10–14]. Rutile is the most stable phase of TiO₂ at ambient conditions, but TiO₂ exhibits a structural phase transition from orthorhombic-I-type to cotunnite-type structures at approximately 50 GPa [11,14]. Cotunnite was previously reported to be the highest pressure phase of TiO₂ [15]. However, a recent *ab initio* study of TiO₂ predicted a transition from the pyrite-type structure to an Fe₂P-type structure (hexagonal, space group *P*62*m*) (Fig. 1) at 690 GPa. This transition bypasses the cotunnite-type phase stability at low temperatures [18]. Unfortunately, because no dioxides or difluorides with the high pressure Fe₂P-type crystal structure have been reported, the physical and chemical properties of the high pressure structural phase of oxides are still unknown.

Studying the high-pressure structural phases of ${\rm TiO}_2$ is key to understanding the rich polymorphism in the metal dioxide systems. ${\rm TiO}_2$ exhibits significantly lower transition pressures to post-cotunnite phases than other dioxides. For instance, the α -PbO $_2$ phase stabilizes at ~10 GPa in ${\rm TiO}_2$, while the same phase stabilizes at 100 GPa in ${\rm SiO}_2$. However, all the studies performed on ${\rm TiO}_2$ have been limited to 100 GPa, and as yet no post-cotunnite phases have been identified. In this study, we theoretically and experimentally investigate the high pressure structural phases of ${\rm TiO}_2$, in particular to determine if this material exhibits a ${\rm Fe}_2{\rm P}$ -type structure at high pressures.

4

The high-pressure and -temperature phase transitions of dioxides are of fundamental interest in solid-state physics, chemistry, and the geosciences. In particular, because of its high refractive index [1-9], TiO₂ is an important wide-gap oxide semiconductor for various industrial applications, such as electrochemical solar cells and photocatalysts. Additionally, high-pressure transformations of TiO₂ have attracted special attention as a low-pressure analog of SiO₂, the most abundant component of the Earth's mantle.

Transition sentences

Previous experimental and theoretical studies reveal many crystalline polymorphs of TiO₂ at high pressures and high temperatures [10–14]. Rutile is the most stable phase of TiO₂ at ambient conditions, but TiO₂ exhibits a structural phase transition from orthorhombic-I-type to cotunnite-type structures at approximately 50 GPa [11,14]. Cotunnite was previously reported to be the highest pressure phase of TiO₂ [15]. However, a recent *ab initio* study of TiO₂ predicted a transition from the pyrite-type structure to an Fe₂P-type structure (hexagonal, space group *P*62*m*) (Fig. 1) at 690 GPa. This transition bypasses the cotunnite-type phase stability at low temperatures [18]. Unfortunately, because no dioxides or difluorides with the high pressure Fe₂P-type crystal structure have been reported, the physical and chemical properties of the high pressure structural phase of oxides are still unknown.

Studying the high-pressure structural phases of TiO_2 is key to understanding the rich polymorphism in the metal dioxide systems. TiO_2 exhibits significantly lower transition pressures to post-cotunnite phases than other dioxides. For instance, the α -PbO $_2$ phase stabilizes at ~10 GPa in TiO_2 , while the same phase stabilizes at 100 GPa in SiO_2 . However, all the studies performed on TiO_2 have been limited to 100 GPa, and as yet no post-cotunnite phases have been identified. In this study, we theoretically and experimentally investigate the high pressure structural phases of TiO_2 , in particular to determine if this material exhibits a Fe_2 P-type structure at high pressures.

5

The high-pressure and -temperature phase transitions of dioxides are of fundamental interest in solid-state physics, chemistry, and the geosciences. For example

Because of its high refractive index [1-9], TiO₂ is a particularly important widegap oxide semiconductor for various industrial applications, such as electrochemical solar cells and photocatalysts. Additionally, high-pressure transformations of TiO₂ have attracted special attention as a low-pressure analog of SiO₂, the most abundant component of the Earth's mantle.

Transition sentences

Previous experimental and theoretical studies reveal many crystalline polymorphs of TiO₂ at high pressures and high temperatures [10–14]. Rutile is the most stable phase of TiO₂ at ambient conditions, but TiO₂ exhibits a structural phase transition from orthorhombic-I-type to cotunnite-type structures at approximately 50 GPa [11,14]. Cotunnite was previously reported to be the highest pressure phase of TiO₂ [15]. However, a recent *ab initio* study of TiO₂ predicted a transition from the pyrite-type structure to an Fe₂P-type structure (hexagonal, space group *P*62*m*) (Fig. 1) at 690 GPa. This transition bypasses the cotunnite-type phase stability at low temperatures [18]. Unfortunately, because no dioxides or difluorides with the high pressure Fe₂P-type crystal structure have been reported, the physical and chemical properties of the high pressure structural phase of oxides are still unknown.

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6